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2082-08-27 Hints & Solution

Section – I

1.(d)	When a particle move with uniform velocity then position of
	particle changes.

- Both wheel travel equal linear distance is same time 2.(c)
- $h = \frac{2T\cos\theta}{\rho gr}$, i.e. pressure 3.(b)
- $\frac{Q}{t} = \frac{KAd\theta}{dl}$ 4.(d)
- For a diabetic process $\Delta Q = 0$ 5.(a) So du + dw = 0 or, -dw = du
- $PE = \frac{1}{4}E_T$ 6.(b) or, $\frac{1}{2} m\omega^2 y^2 = \frac{1}{4} \times \frac{1}{2} m\omega^2 (A^2)$ or, $y = \frac{A}{2}$
- At Q, E = $\frac{2Q}{4\pi\epsilon_0 r^2}$ 7.(d)
 - At -2Q, E' = $\frac{Q}{4\pi\epsilon_0 r^2} = \frac{E}{2}$
- 8.(d)
 $$\begin{split} R_{eq} &= \frac{R_{ABC} \times R_{AC}}{R_{ABC} + R_{AC}} = \frac{8 \times 4}{8 + 4} = \frac{3}{4} \\ M &= \sqrt{M^2 + M^2 + 2M^2 \cos 60} \end{split}$$
- 9.(c)
- 10.(c)
- 11.(d)

f will be maximum if μ is least

Where
$$\mu = A + \frac{B}{\lambda^2}$$
 i.e. $\lambda_r > \lambda_v$

- $So, ~~\mu_r \leq \mu_v$ $\frac{\beta_r}{\beta_v} = \frac{\lambda_r}{\lambda_v} = \frac{8000 \times 10^{-10}}{4000 \times 10^{-10}} = 2:1$ 12.(b)
- $mvr = \frac{nh}{2\pi}$ 13.(a) or, $mv = \frac{nh}{2\pi r}$
 - $\therefore \quad \lambda = \frac{h}{m \, v} = \frac{h}{nh} \times 2\pi r = \frac{2\pi r}{n}$
- $\alpha = 0.98 = \frac{I_c}{I}$ 14.(a)
 - $\beta = \frac{\alpha}{1 \alpha} = \frac{0.98}{1 0.98} = 49$
- Last electron of Cl is 3P 15.(b) $n = 3, l = 1, m = 0, s = -\frac{1}{2}$
- 16.(c) Al Mg

p	q	11	12	13
e	10	10	10	10

Most force of attraction between electron & Nucleus in case Al so smaller size for Al⁺⁺⁺ than Mg⁺⁺ than Na⁺ than F⁻.

- $\Delta n = 2 1 = 1$ 17.(b) $\therefore K_p = K_c (RT)$ $\frac{K_p}{K_c} = RT$ $K_p > K_c$
- Since concⁿ B is double rate 4 times thus order of B 18.(b) is 2nd so, A is 1st.
- 19.(b) Only Mg and Mn can displace hydrogen from dil. HNO3.
- 20.(d)
- $NH_3 + CO_2 + H_2O \rightarrow NH_4HCO_3$ 21.(b) $NH_4HCO_3 + NaCl \rightarrow NaHCO_3 \downarrow + NH_4Cl$ White crystal
- 2s² 2p⁶ 3s² 2p⁶ 3s² 3p⁶ 3d⁵ (most 22.(b)
- $NaCNS + FeCl_3 \rightarrow Fe(CNS)_3 + NaCl$ 23.(d)
- blood red Priority of numbering goes to double bond.
- 24.(d) 25.(b)
- Reimer Tieman reaction) 26.(b)
- 27.(d) 28.(c)
- Given equations $x^2 + bx + c = 0$, $x^2 + dx + e = 0$ 29.(c)

$$x^2 + dx + e = 0$$

If the both roots are common then
$$\frac{1}{1} + \frac{b}{d} = \frac{c}{e} \Rightarrow be = cd$$

Given $1 + 2 + 3 + ... + n = \frac{1}{5} (1^2 + 2^2 + ... + n^2)$ 30.(c)

or,
$$\frac{n(n+1)}{2} = \frac{1}{5} \frac{n(n+1)(2n+1)}{6}$$

- 2n + 1 = 15
- n = 7i.e.
- Using properties, $(AB)^T = B^T A^T$ 31.(d)
- 32.(b) $A \cap B = \{3, 4\}, n(A \cap B) = 2$ So $n[(A \times B) \cap (B \times A)] = \{n(A \cap B)\}^2 = 2^2 = 4$
- By definition, projection of \vec{b} on $\vec{a} = \frac{\vec{a} \cdot \vec{b}}{\vec{c}}$ 33.(a)
- 34.(a)
- $x^{2} + y^{2} = a^{2}\cos^{2}\theta + a^{2}\sin^{2}\theta = a^{2}$ Circle is $x^{2} + y^{2} 4x 6y 12 = 0$ 35.(d) Comparing with $x^2 + y^2 + 2gx + 2fy + c = 0$ g = -2, f = -3, c = -12So, $r = \sqrt{g^2 + f^2 - c} = 5$

$$g = -2$$
, $f = -3$, $c = -12$
So, $r = \sqrt{g^2 + f^2 - c} = 5$

- and circumference = $2\pi r = 2\pi \times 5 = 10\pi$
- 36.(d)
- Eqⁿ is $y^2 = 4ax + 4a^2$ i.e. $(y-0)^2 = 4a\{x-(-a)\}$
 - So the vertex is (-a, 0)
- $\sin^2 \alpha + \sin^2 \beta + \sin^2 \gamma = 1 \cos^2 \alpha + 1 \cos^2 \beta + 1 \cos^2 \beta$ 37.(b) $= 3 - (\cos^2 \alpha + \cos^2 \beta + \cos^2 \gamma) = 3 - 1 = 2$
- 38.(c) $\sin^{-1}x + \cos^{-1}x = \frac{\pi}{2}$
 - i.e. $\cos^{-1} x = \frac{\pi}{2} \sin^{-1} x = \frac{\pi}{2} \frac{\pi}{10} = \frac{5\pi \pi}{10}$

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$$=\frac{4\pi}{10} = \frac{2\pi}{5}$$
39.(b) $Z^2 = (1+i)^2 = 1 + 2i + i^2 = 1 + 2i - 1 = 2i$
So multiplicative inverse of
$$Z^2 = \frac{1}{2i} - \frac{i^2}{2i} = -\frac{i}{2}$$
40.(c) Fact
41.(b) $P(B) = 1 - P(B) = 1 - \frac{5}{8} = \frac{3}{8}$
and $P(A/B) = \frac{P(A \cap B)}{P(B)} = \frac{\frac{1}{4}}{\frac{3}{8}} = \frac{1}{4} \times \frac{8}{3} = \frac{2}{3}$
42.(a) $\frac{2}{2!} + \frac{3}{3!} + \frac{4}{4!} + ...$

$$= \frac{1}{1!} + \frac{1}{2!} + \frac{1}{3!} + ... = 1 + \frac{1}{1!} + \frac{1}{2!} + \frac{1}{3!} + ... - 1$$

$$= e - \frac{1}{43.(c)} \quad \lim_{x \to 0^+} \frac{e^x}{1} = \lim_{x \to 0^+} \frac{1}{1 + e^{1/a}} = \frac{1}{1 + 0} = 1$$

$$e^x + 1$$
44.(b) $y = \sin^{-1}(3x - 4x^3) = 3\sin^{-1}x$
So, $\frac{dy}{dx} = \frac{3}{\sqrt{1 - x^2}}$
45.(d) $I = \int_{\sin^4 x} \frac{\sin^4 x}{3x} dx = \int_{\cos^4 x} \frac{\sin^4 x}{3\cos^2 x} dx$

$$= \int_{\tan^4 x} \sec^2 x dx = \frac{\tan^5 x}{5} + c$$
46.(b) For tangent parallel to x-axis, put $x = 0$, $y^2 = 25$
47.(b) $I.F. = e^{i\frac{2}{3}} = \frac{3}{5} \sec x$

$$\Rightarrow IPdx = \log_5 \sec x \Rightarrow IPdx$$

 $= \frac{3}{2} \, \text{mR}^2 = \frac{3}{2} \times 2 \times 0.1^2$

 $T_1 = \frac{YA(l_1 - l)}{l}$

 $=0.03 \text{ kgm}^2$

1st case

63.(c)

or,
$$\frac{T_1l}{l_1-l} = YA \dots$$
 (i)

 2^{nd} case

 $T_2 = \frac{YA(l_2-l)}{l}$

or, $\frac{T_2l}{l_2-l} = YA \dots$ (2)

From (1) & (2)

 $\frac{T_1l}{l_1-l} = \frac{T_2l}{l_2-l}$

or, $T_1l = T_2l = T_2l$

or, $I = \frac{T_1l_2-T_2l_1}{T_1-T_2}$

64.(a) m be mass of steam condensed m × 540 + m(100 – 90) = 22 × (90 – 20)

or, $m = \frac{22 \times 70}{560} = 2.75g$

Water $122 + 2.75 = 24.75g$

65.(c)

 $\frac{T_2}{T_1} = \frac{T_2l}{T_1} = \frac{T_2l}{T_2} = \frac{$

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or,
$$\frac{1}{v} = \frac{(m+1)}{mf} \implies v = -\frac{mf}{m+1}$$

$$\therefore \quad \text{Magnification} = \frac{v}{u} = \frac{mf}{(m+1) \times mf}$$

$$= \frac{1}{m+1}$$

71.(d)
$$\frac{x}{D} = \frac{\lambda}{d}$$

 $x = \frac{D\lambda}{d}$
 $\therefore 2x = \frac{2D\lambda}{d} = \frac{2 \times 2 \times 600 \times 10^{-9}}{10^{-3}}$
 $= 2.4 \times 10^{-3} \text{ m}$
 $= 2.4 \text{ mm}$

72.(d)
$$1^{st} case \frac{hc}{\lambda} = \phi + \frac{1}{2} mv^{2}$$
or, $v = \sqrt{\frac{2}{m} \left(\frac{hc}{\lambda} - \phi\right)}$

$$2^{nd} case \frac{4hc}{3\lambda} = \phi + \frac{1}{2} mv^{2}$$
or, $v' = \sqrt{\frac{2}{3} \left(\frac{4hc}{3\lambda} - \phi\right)}$

$$\therefore \frac{v'}{v} = \sqrt{\frac{\frac{4}{3} \frac{hc}{\lambda} - \phi}{\frac{hc}{\lambda} - \phi}}$$

73.(a)
$$\frac{A}{A_0} = \left(\frac{1}{2}\right)^{\frac{1}{1_{1/2}}}$$
or,
$$A = A_0 \left(\frac{1}{2}\right)^{\frac{30}{0.693} \times \frac{6.93}{60}}$$

$$= 128 \left(\frac{1}{2}\right)^{\frac{6.93}{0.693 \times 2}}$$

ratio (1:2:3) and mole ratio in the 74.(d) Given mole reaction (2:6:3)

Thus, H2O is limiting reagent

6 mole H₂O gives 4 mole Fe(OH)₃ 1 mole H_2O gives $\frac{4}{6}$ mole $Fe(OH)_3$ 2 mole H_2O gives $\frac{4}{6} \times 2$ mole Fe(OH)₃

1.34 mole Fe(OH)₃

Since 3 mole of $Zn(NO_3)_2 \Rightarrow 6HNO_3$ as acid 75.(c) 2 mole NO is formed by reduction of HNO₃ 2 mole HNO₃ act as oxidizing agent.

76.(a)
$$T = 273$$
 $P_T = 1$ atm
 $R = 0.0821$
 $nH_2 = \frac{4}{2} = 2$ mole & $nN_2 = \frac{2.8}{28} = 0.1$
 $V_T = \frac{n_T RT}{P_T} = \frac{2.1 \times 0.0821 \times 273}{1}$
 $= 47.07$ litre

$$W = \frac{EVN}{1000} = 7.17 \text{ gm}$$

78.(b) 79.(b)

80.(a) Both HCl & NaCl are strong electrolyte.

81.(c) $4KI + 2CuSO_4 \rightarrow Cu_2I_2\downarrow + 2K_2SO_4 + I_2$

82.(c) As $1 + x^2 \neq 0$ for real x.

So domain = R $x^2 \ge 0$ so $1 + x^2 > x^2$ So, $\frac{x^2}{1 + x^2} < 1$ i.e. y < 1

 $\begin{array}{ll} \therefore & Range = [0,1) \\ We \ have \ (1+x)^n = C_0 + C_1 x + C_2 x^2 + ... + C_n x^n \\ Integrating, \ \frac{(1+x)^{n+1}}{n+1} = C_0 x + \frac{C_1}{2} \ x^2 + \frac{C_2}{3} \ x^3 + + C_n x^n \end{array}$ 83.(b)

$$\frac{C_n}{n+1} x^{n+1} + K$$

Putting x = 0, $K = \frac{1}{n+1}$ Then $\frac{(1+x)^{n+1}}{1+1} = C_0x + \frac{C_1}{2}x^2 + \frac{C_2}{3}x^3 + \dots + \frac{C_n}{n+1}$ $x^{n+1} + \frac{1}{n+1}$ Putting x = 1, we get $C_1 + \frac{C_2}{2} + \dots$

84.(c)

Given word is SOCIETY There are 3 vowels and 4 consonants so there are 3 choices for vowels and 4 choices for consonants for the alternate arrangements so total no. of arrangements = $P(3,3) \times P(4,4)$ = 3! \times 4!

85.(b)
$$x + \frac{1}{x} = -1$$
 i.e. $x^2 + x + 1 = 0$

i.e.
$$x = \frac{-1 \pm \sqrt{1^2 - 4.1.1}}{2.1} = \frac{-1 \pm \sqrt{3}i}{2} = w, w^2$$

So, $x^{2010} + \frac{1}{x^{2010}} = w^{2010} + \frac{1}{w^{2010}}$
 $= (w^3)^{670} + \frac{1}{(w^3)670} = 1 + \frac{1}{1} = 2$

86.(c) Given
$$|\vec{a}| = |\vec{b}| = |\vec{c}| = 1$$

and
$$\vec{a} + \vec{b} + \vec{c} = \vec{0}$$

So,
$$\vec{b} + \vec{c} = -\vec{a}$$

$$(\vec{b} + \vec{c})^2 = (-\vec{a})^2$$

or,
$$b^2 + c^2 + 2|\vec{b}| |\vec{c}| \cos\theta = a^2$$

or,
$$1 + 1 + 2\cos\theta = 1$$

or, $2\cos\theta = -1$

or,
$$2\cos\theta = -1$$

or,
$$\cos\theta = -\frac{1}{2}$$
 i.e. $\theta = 120^{\circ}$

$$ax^{2} + 2hxy - ay^{2} = 0$$
 ...(i)
 $bx^{2} + 2gxy - by^{2} = 0$...(ii)

Given equations are $ax^2 + 2hxy - ay^2 = 0 ...(i)$ $bx^2 + 2gxy - by^2 = 0 ...(ii)$ Eqⁿ of bisector of angle between lines represented by

(1) Is
$$h(x^2 - y^2) = (a + a) xy$$

i.e. $hx^2 - 2axy - hy^2 = 0$ (iii)
Since (ii) & (iii) are identical,
So, $\frac{b}{h} = \frac{2g}{-2a} = \frac{-b}{h}$

So,
$$\frac{b}{h} = \frac{2g}{-2a} = \frac{-b}{h}$$

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88.(c)
$$(c) = \frac{1}{2} + \frac{1}{2} +$$

...The End...