

Section - I

- 1.(b) 2.(c) 3.(a) 4.(d) 5.(d) 6.(c)
7.(c) 8.(a) 9.(b) 10.(d) 11.(d) 12.(b)
- 13.(a) Ethyne (terminal alkyne) forms copper acetylide precipitate whereas ethane does not.
- 14.(a) Because $\text{HCO}_3^- \rightarrow \text{CO}_3^{--} + \text{H}^+$ & $\text{HCO}_3^- + \text{H}^+ \rightarrow \text{H}_2\text{CO}_3$.
Here both CO_3^{--} & H_2CO_3 are stable.
- 15.(b) $\text{SO}_2 + 2\text{H}_2\text{O} \rightarrow \text{H}_2\text{SO}_4 + 2[\text{H}]$
The nascent hydrogen turns colored substance to colorless compounds.
- 16.(c) Fluorine is most electronegative element.
 $F = 1s^2 2s^2 2p^5$
- 17.(a) Because CaO absorbs moisture without reacting with NH_3 .
- 18.(d) High pressure shifts rxn towards low volume.
Exothermic reaction favors low temperature.
- 19.(d) NaOH is commercially manufactured by electrolysis of brine.
- 20.(c) Because both have same molecular formula but different functional groups $-\text{COOH}$ & ester respectively.
- 21.(c) $\text{O}_2 + \text{C} \rightarrow \text{CO}_2$
1 mole C \rightarrow 1 mole O_2
and 1 mole at NTP = 22.4 litres
- 22.(b) Magnetic quantum number specifies orbital orientation.
- 23.(a) Malachite = $\text{CuCO}_3 \cdot \text{Cu}(\text{OH})_2$
- 24.(c) $\text{CH}_3\text{CH}_2\text{I} + \text{CH}_3\text{ONa} \rightarrow \text{CH}_3\text{OCH}_2\text{CH}_3 + \text{NaI}$
Williamson's synthesis gives ether from sodium alkoxide.
- 25.(a) Sugar (sucrose) has molecular weight of 342 g/mol.
- 26.(b) Electron affinity for Halogens: $\text{Cl} > \text{F} > \text{Br} > \text{I}$
- 27.(c) $\frac{\text{Nucleon}}{\text{neutron}} = \frac{235}{235 - 92} = \frac{235}{143}$
- 28.(a) $v = \sqrt{\frac{E}{\rho}}$, ratio of elasticity & density of solid is maximum even density & elasticity is max.
- 29.(a) Internal energy changes only when temperature change.

- 30.(c) $R = H$
or, $\frac{u^2 \sin 2\theta}{g} = \frac{u^2 \sin^2 \theta}{2g}$
or, $2 \sin \theta \cdot \cos \theta = \frac{\sin^2 \theta}{2}$
or, $\tan \theta = 4$
or, $\theta = \tan^{-1}(4)$
- 31.(a) Bob with half filled with Hg will have CG lower so period of oscillation will be more.
- 32.(c) $\beta = \frac{D\lambda}{d}$
 $d' = \frac{d}{2}$ & $D' = 2D$ so
 $\beta' = \frac{D'\lambda}{d'} = \frac{2D\lambda}{d} \times 2 = 4\beta$
- 33.(c) $\frac{R'}{R} = \left(\frac{2l'}{l}\right)^2 = 4$
 $R' = 4R$
- 34.(a) Lose in time in a day = $\frac{1}{2} \propto \Delta \theta \times 1 \text{ day}$
 $= \frac{1}{2} \propto \theta \times 86400 \text{ sec}$
 $= 43200 \propto \theta \text{ sec}$
- 35.(d) $\beta = \frac{\alpha}{1 - \alpha} = \frac{0.99}{1 - 0.99} = \frac{0.99}{0.01}$
- 36.(c) When switch is made on then net resistance decreases so current increases due to which brightness of X increases and current in Y decreases ie brightness decreases.
- 37.(c) $\text{KE} = \frac{hc}{\lambda} - \phi$
 $= \frac{6.62 \times 10^{-34} \times 3 \times 10^8}{300 \times 10^{-19}} - 1.6 \times 10^{-19} - 2.46$
 $= 1.67 \text{ eV}$
- 38.(a) **First case**
 $R = \sqrt{P^2 + 2PQ\cos\theta + Q^2} \dots (i)$
2nd case
 $\tan 90^\circ = \frac{2Q\sin\theta}{P + 2Q\cos\theta} = \frac{1}{0}$
or, $P + 2Q\cos\theta = 0$

- or, $\cos\theta = -\frac{P}{2Q}$ (ii)
- From (i) & (ii)
- $$R = \sqrt{P^2 + 2PQ \left(-\frac{P}{2Q}\right) + Q^2} = Q$$
- 39.(c) $\frac{t_2}{t_1} = \frac{\sqrt{\frac{h}{2} - 0}}{\sqrt{h - 0}} = \frac{1}{\sqrt{2}}$
- $\therefore t_2 = \frac{10}{\sqrt{2}} = 7 \text{ min}$
- 40.(d) $3f_0^c = 2f_0^0$
- or, $3\frac{v}{4l_c} = 2 \times \frac{v}{2l_0}$
- or, $\frac{l_c}{l_0} = \frac{3}{4}$
- 41.(c) $A^{-1} = \frac{1}{|A|} \text{adj } A$; $\text{adj } A = \begin{bmatrix} d & -b \\ -c & a \end{bmatrix}$
- 42.(b) Gradient means slope.
Slope = $\tan\theta$ (θ = angle with +ve x-axis)
- 43.(b) Any vector in 3D is represented using \hat{i} , \hat{j} and \hat{k} .
- 44.(c) $f'(x) = \frac{d}{dx} \left(\frac{1}{x}\right) = -\frac{1}{x^2}$
- 45.(c) Because $\cos(-\theta) = \cos\theta$ and \cos is +ve in 1st and 4th Quadrant. i.e. 2π (turnaround)
- 46.(c) A function can't have multiple values at same point (x).
- 47.(d) Think of it as $(x, y) = (x, y, 0)$
i.e. $x\hat{i} + y\hat{j} = x\hat{i} + y\hat{j} + 0\hat{k}$
- 48.(b) Put $x = t^2$ and solve.
- 49.(b)
- 50.(b) Parabola (degree 2) has 1 turning point.
St. line (degree 1) has no turning point.
- 51.(c) $P(5H, 0T) + P(4H, 1T) + P(3H, 2T)$
 $P(aH, bT) = {}^{a+b}C_a \{P(H)\}^a \times \{P(T)\}^b$
- 52.(b)
- 53.(a) Put $x = 45^\circ$, $\tan x + \cot x = 2$ which is minimum among option.
- 54.(d) Functions are on-to-one or many to-one relations.
Each element in A has 4 possible mappings.

- 55.(c) $\log_x 125 = 3$
 $\Rightarrow 125 = x^3$
- 56.(c)
- 57.(c) $d = \left| \frac{d_1 - d_2}{\sqrt{a^2 + b^2 + c^2}} \right|$ both eqns must have same a, b & c.
- 58.(b)
- 59.(c) $f(x) = e^x$
 $\Rightarrow f^{-1}(x) = \log_e x$
- 60.(c) Standard relation.

Section – II

- 61.(c) 62.(b) 63.(d) 64.(a)
- 65.(b) $\text{NaCl} + \text{H}_2\text{O} \rightarrow \text{Na}^+ + \text{Cl}^- + \text{H}^+ + \text{OH}^-$
Reduction potential of $\text{Na}^+ < \text{H}^+$
Oxidation potential of $\text{Cl}^- > \text{OH}^-$
- 66.(a)
- $$2\text{HNO}_3 \xrightarrow{\text{P}_2\text{O}_5} \text{N}_2\text{O}_5 + \text{H}_2\text{O}$$
- P_2O_5 acts as dehydrating agent
- 67.(a) Reaction of haloalkane with KCN increases a carbon atom in haloalkane's product.
- 68.(c) $[\text{H}^+] = [\text{H}_3\text{O}^+]$
 $\text{pH} = -\log[\text{H}^+]$
- 69.(a) Eq. base = $\frac{W}{E} \times 100$
Eq. acid = $N \times V_m$
- 70.(d) $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
So, 1 mole CaCO_3 gives 1 mole CO_2
- 71.(d) $\text{NH}_4\text{Cl} + \text{NaNO}_2 \rightarrow \text{NH}_4\text{NO}_2 + \text{NaCl}$
 $\text{NH}_4\text{NO}_2 \xrightarrow{\Delta} \text{N}_2 + 2\text{H}_2\text{O}$
- 72.(a)
- 73.(c)

For 1st part

$$C' = \frac{12 \times 24}{12 + 24} = \frac{12 \times 24}{36} = 8 \mu\text{F}$$

Across ac

$$C'' = 8 + 6 + 4 = 18 \mu\text{F}$$

Across ab

$$C_{\text{eq}} = \frac{18 \times 9}{18 + 9} = \frac{18 \times 9}{27} = 6 \mu\text{F}$$

74.(b) $\tan 45^\circ = \frac{EQ}{mg}$
 or, $EQ = mg$
 or, $\frac{V}{d} Q = mg$
 or, $Q = \frac{mgd}{V} = \frac{1.92 \times 10^{-14} \times 10 \times 5 \times 10^{-3}}{3000}$
 $= 3.2 \times 10^{-19} \text{C}$
 $= \frac{3.2 \times 10^{-19}}{1.6 \times 10^{-19}} e$
 $= 2e$

75.(a) Change in wt = change in upthrust
 or, $300 = l^2 \times 3$
 or, $R^2 = 100$
 or, $l = 10 \text{ cm}$

76.(a) $R_{eq} = \frac{x(20-x)}{20-x+x}$
 or, $1.8 \times 20 = 20x - x^2$
 or, $x^2 - 20x + 36 = 0$
 or, $x - 2x - 18x + 36 = 0$
 $x = 2\Omega \text{ \& } 18\Omega$

Shorter is 2Ω
 $\therefore \text{ length} = 2\text{m}$

77.(d) $B_p = 4B_Q$
 or, $\mu_0 n_1 I_1 = 4\mu_0 n_2 I_2$
 or, $\frac{I_1}{I_2} = \frac{4n_2}{n_1}$

78.(b) $5\% \text{ of } p = \frac{n hc}{t \lambda}$
 or, $\frac{n}{t} = \frac{5}{100} \times 100 \times \frac{\lambda}{hc}$
 $= \frac{5 \times 555 \times 10^{-9}}{6.62 \times 10^{-34} \times 3 \times 10^8}$
 $= 13.9 \times 10^{18}$

79.(c) Energy required (E) = $4\pi R^2 T(n^{1/3} - 1)$
 $= \pi d^2 T(64^{1/3} - 1)$
 $= 3\pi d^2 T$

80.(a) $0^2 = I_1 I_2 = 6 \times \frac{2}{3}$
 or, $0 = 2\text{cm}$

81.(a) $E_2 - E_1 = hf$
 or, $f = \frac{\{-3.4 - (-13.6)\} \times 1.6 \times 10^{-19}}{6.62 \times 10^{-34}}$
 $= 2.46 \times 10^{15} \text{ Hz}$

82.(d) For glass air interface
 $\sin C = \frac{1}{\mu} = \frac{1}{1.5}$
 $C = \sin^{-1} \left(\frac{1}{1.5} \right) = 41.8^\circ$
 $A = r_1 + C$
 or, $r_1 = 60 - 41.8 = 18.2^\circ$

At Q, $\mu = \frac{\sin i}{\sin r_1}$
 or, $i = \sin^{-1}(1.5 \times \sin 18.2) = 28^\circ$

83.(a) $\tan \theta = \frac{F}{m(g+a)}$
 or, $\tan 45^\circ > \frac{F}{m(g+a)}$
 or, $F = 2(10+2) = 24\text{N}$

84.(c) $T_2 V_2^{r-1} = T_1 V_1^{r-1}$
 or, $T_2 = 300 \left(\frac{V_1}{V_2} \right)^{1.4-1}$
 $= 300(9)^{0.4} = 722\text{K} = 449.5^\circ$

85.(b) $I_{max} = \frac{V}{P} = \frac{120}{15} = 8\text{A}$
 $I = 80\% \text{ of } I_{max} = \frac{80}{100} \times 8 = 6.4 \text{ A}$
 Now, $V - L \frac{dI}{dt} = IR$

or, $L \frac{dI}{dt} = 120 - 6.4 \times 15$

or, $\frac{dI}{dt} = \frac{24}{0.6} = 40 \text{ A/s}$

86.(b) Put $n = 1$, expression becomes $1 + 3^1 c_1$

87.(b) Eqn of ellipse with centre (h, k) is
 $\frac{(x-h)^2}{a^2} + \frac{(y-k)^2}{b^2} = 1$

Directrix is parallel to minor axis.

88.(c) Differentiate wrt x on both sides and simplify.

89.(c) Circle is: $x^2 + y^2 + 2gx + 2fy + c = 0$
 Centre = $(-g, -f)$

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2083-2-30 (SET-B) Hints & Solution

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| <p>90.(c) $\frac{dA}{dt} = \frac{d}{dt} \frac{\sqrt{3}}{4} a^2 = \frac{\sqrt{3}}{2} a \frac{da}{dt}$</p> <p>91.(b) $\int \frac{dx}{\sqrt{1-x^2}} = \sin^{-1}x$ (standard)</p> <p>92.(d) By remainder theorem
 $K(-2)^3 + 3(-2)^2 - 13 = 2(-2)^3 - 5(-2) + K$</p> <p>93.(d) Break into 2 determinants, then simplify and take common determinant.
 i.e. $a \det - b \det = (a - b) \det$.</p> <p>94.(c) Area of parallelogram = $\vec{a} \times \vec{b}$</p> | <p>95.(b) Shortest way is to solve using hit and trial method.</p> <p>96.(b)</p> <p>97.(d) Use calculator</p> <p>98.(c) $3z_1 - 4z_2 = -6 + 11i$
 $z = \sqrt{x^2 + y^2}$</p> <p>99.(b) Put $n = 2$</p> <p>100.(c) $x^2 = 4ay \Rightarrow a = 3$
 Focus = $(h, k + a) = (2, 3 - 3) = (2, 0)$</p> |
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...Best of Luck...