

Section – I

1.(a) $\vec{P} \times \vec{Q} = |\vec{P}| |\vec{Q}| \sin \hat{n} = 0$

$\therefore \sin \theta = \sin 0^\circ$

or, $\theta = 0^\circ$

$R = \sqrt{P^2 + 2PQ \cos 0^\circ + Q^2} = \sqrt{(P+Q)^2} = P+Q$

2.(c) When body crosses top most position with critical velocity the velocity at horizontal position is $v = \sqrt{3gr}$

$a = \frac{v^2}{r} = \frac{3gr}{r} = 3g$

3.(b)
$$\frac{v_{c1}}{v_{c2}} = \frac{\sqrt{\frac{2GM_1}{R_1}}}{\sqrt{\frac{2GM_2}{R_2}}} = \sqrt{\frac{M_1}{R_1} \times \frac{R_2}{M_2}}$$

$$= \sqrt{\frac{4\pi R^2 \rho}{3} \times \frac{2R}{4\pi (2R)^3 \rho}}$$

$$= \frac{R^2}{4R^2} = \frac{1}{2}$$

4.(c) $T = \text{stress} \times A = m r \omega^2$

or, $\omega = \sqrt{\frac{\text{Stress} \times A}{m l}}$

$$= \sqrt{\frac{4.8 \times 10^7 \times 10^{-6}}{10 \times 0.3}} = 4 \text{ rad/s}$$

5.(c) $\omega t_1 = \omega t_2$

Again $\omega t_1 - (\text{upthrust})_1 = \omega t_2 - (\text{upthrust})_2$

or, $(\text{upthrust})_1 = (\text{upthrust})_2$

or, $2 \times \rho_0 g = V \rho g$

or, $\rho_l = 2 \times 0.9 = 1.8 \text{ g/cc}$

6.(b) $\alpha = \frac{\Delta l}{l \Delta \theta}$

or, $\Delta \theta = \frac{1.1 \times 10^{-3}}{10 \times 1.1 \times 10^{-3}} = 10^\circ \text{C}$

7.(b) $VP^2 = \text{constant}$

or, $V \left(\frac{nRT}{V} \right)^2 = \text{constant}$

or, $\frac{T^2}{V} = \text{constant}$

or, $\frac{T^2}{2V} = \frac{T_0^2}{V}$

or, $T = \sqrt{2} T_0$

8.(d) $f_0^c = 3f_0^c$

or, $\frac{v}{2l_0} = 3 \times \frac{v}{4l_c}$

or, $\frac{l_0}{l_c} = \frac{2}{3}$

9.(c) $\frac{R'}{R} = \left(\frac{d}{d'} \right)^4 = \left(\frac{d}{\frac{d}{3}} \right)^4 = 81$

$R' = 81R$

10.(a) $F = B_1 I l$

$= \frac{\mu_0 I}{2\pi a} I l$

$= \frac{4\pi \times 10^{-7} \times 1 \times 1 \times 0.5}{2\pi \times 1}$

$= 10^{-7} \text{ N}$

11.(a) $I = \frac{V_{rms}}{X_L} = \frac{V_0}{\sqrt{2} \times \omega L}$

$= \frac{282}{\sqrt{2} \times 314 \times 0.1}$

$= 6.35 \text{ A}$

12.(b) $I = \frac{ne}{t}$

$\frac{n}{t} = \frac{I}{e} = \frac{4.8 \times 10^{-3}}{1.6 \times 10^{-19}}$

$= 3 \times 10^{16}$

13.(b) $\lambda = \frac{h}{mv} = \frac{6.62 \times 10^{-34}}{9.1 \times 10^{-31} \times 2 \times 10^6}$

$= 3.64 \times 10^{-10} \text{ m}$

$= 3.64 \text{ \AA}$

14.(c) $R = \frac{\Delta V}{\Delta I} = \frac{0.7 - 0.5}{10^{-3}} = 200 \Omega$

15.(c) **Hint:** One molecule of CO_2 contains 3 atoms (1 C + 2 O). Multiply by Avogadro's number.

Solution: 1 molecule of CO_2 has 3 atoms.

Total atoms $= 3 \times 6.022 \times 10^{23} = 1.806 \times 10^{24}$.

16.(b) **Hint:** Find the molar mass of NH_4NO_3 , then take the mass of the two nitrogen atoms as a percentage.

Solution: Molar mass of $\text{NH}_4\text{NO}_3 = 14 + 4 + 14 + 48 = 80$.

Mass of N $= 2 \times 14 = 28$.

%N $= (28/80) \times 100 = 35\%$.

17.(b) **Hint:** This is Gay-Lussac's (pressure) law. Pressure and temperature must both be measured on the absolute (Kelvin) scale.

Solution: By Gay-Lussac's law, at constant volume $P \propto T$, where T is the absolute (Kelvin) temperature.

18.(c) **Hint:** The number of orbitals in a shell equals n^2 . (The maximum number of electrons would be $2n^2$.)

Solution: Number of orbitals $= n^2 = 4^2 = 16$. (These 16 orbitals can hold a maximum of $2n^2 = 32$ electrons.)

19.(d) **Hint:** All four belong to the same group (alkali metals). Atomic radius increases down a group.

Solution: Down group 1, atomic radius increases (more shells). Rb is lowest in the group, so it has the largest atomic radius.

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2083-3-27 Hints & Solution

- 20.(d)** **Hint:** Take oxygen as -2 and remember the overall charge on the ion is -1. Solve for Mn.
Solution: Let oxidation number of Mn = x. $x + 4(-2) = -1 \Rightarrow x - 8 = -1 \Rightarrow x = +7$.
- 21.(a)** **Hint:** It is the salt of a strong acid (HCl) and a weak base (NH₄OH). Consider cationic hydrolysis.
Solution: NH₄Cl is the salt of a strong acid and a weak base. The NH₄⁺ ion hydrolyses to give H₃O⁺, so the solution is acidic (pH < 7).
- 22.(c)** **Hint:** At the cathode, the ion with the higher (less negative) reduction potential is discharged. Compare Cu²⁺ with H⁺.
Solution: Cu²⁺ is more easily reduced than H⁺ (higher reduction potential), so copper metal is deposited at the cathode: $\text{Cu}^{2+} + 2\text{e}^- \rightarrow \text{Cu}$.
- 23.(a)** **Hint:** CO₂ is a linear molecule (O = C = O). The central carbon has two sigma bonds and no lone pairs.
Solution: In CO₂ (O = C = O), carbon forms 2 sigma bonds with no lone pair (2 electron domains) → sp hybridisation, giving a linear shape.
- 24.(a)** **Hint:** In an endothermic reaction, heat is absorbed, so the products have more enthalpy than the reactants.
Solution: An endothermic reaction absorbs heat, so the enthalpy of products > enthalpy of reactants, giving ΔH positive (+).
- 25.(c)** **Hint:** The gas must be chemically unreactive so it does not react with the hot filament. Look among the noble gases.
Solution: Argon, a noble (inert) gas, is used to fill electric bulbs because it does not react with the hot tungsten filament.
- 26.(b)** **Hint:** The most abundant element by mass in the crust is a non-metal present in most rocks and minerals (silicates, oxides).
Solution: Oxygen is the most abundant element by mass in the Earth's crust (≈ 46%), followed by silicon.
- 27.(b)** **Hint:** The coinage metals are the three Group 11 elements traditionally used to make coins.
Solution: The coinage metals are copper, silver and gold (Group 11). Hence copper is the coinage metal among the options.
- 28.(b)** **Hint:** Distinguish this from roasting (heating in a plentiful supply of air). Here air is largely excluded.
Solution: Calcination is the strong heating of an ore in the absence or limited supply of air, driving off volatile matter (e.g. moisture, CO₂). (Roasting is done in excess air.)
- 29.(c)** Given function is $f(x) = \sin \pi x$
 So, period = $\frac{2\pi}{\pi} = 2$
- 30.(b)** Given $f(x) = 10^x + 10^{-x}$
 So, $f(-x) = 10^{-x} + 10^{(-x)} = 10^{-x} + 10^x = 10^x + 10^{-x} = f(x)$
- 31.(b)** ∴ $f(x)$ is even
 Given equation is $x^2 - 5x + 4 = 0$
 Here, discriminant = $(-5)^2 - 4 \cdot 1 \cdot 4 = 25 - 16 = 9$ which is perfect square.
 ∴ Roots are rational and unequal.
- 32.(d)** As the matrix $\begin{pmatrix} 6 & x-2 \\ 3 & x \end{pmatrix}$ has no inverse
 So $\begin{vmatrix} 6 & x-2 \\ 3 & x \end{vmatrix} = 0$
 or, $6x - 3(x-2) = 0$
 or, $6x - 3x + 6 = 0$
 or, $3x = -6 \quad \therefore x = -2$
- 33.(c)** Here no. of points = 10 no. of collinear points = 4
 ∴ No. of triangles = $C(10, 3) - C(4, 3) = 116$
- 34.(b)** $\sin^{-1}x + \sin^{-1}y = \frac{2\pi}{3}$
 or, $\frac{\pi}{2} - \cos^{-1}x + \frac{\pi}{2} - \cos^{-1}y = \frac{2\pi}{3}$
 or, $\cos^{-1}x + \cos^{-1}y = \pi - \frac{2\pi}{3}$
 ∴ $\cos^{-1}x + \cos^{-1}y = \frac{\pi}{3}$
- 35.(c)** $\vec{i} \cdot (\vec{j} \times \vec{k}) + \vec{j} \cdot (\vec{k} \times \vec{i}) + \vec{k} \cdot (\vec{i} \times \vec{j})$
 $= \vec{i} \cdot \vec{i} + \vec{j} \cdot \vec{j} + \vec{k} \cdot \vec{k}$
 $= 1 + 1 + 1 = 3$
- 36.(a)** Given equation is $kx^2 - 4xy + 5y^2 = 0$
 As the lines are coincident so $(-2)^2 = k \times 5$
 i.e. $k = \frac{4}{5}$
- 37.(c)** Here $\frac{1}{2}, \frac{1}{3}, k$ are direction cosines
 So, $\left(\frac{1}{2}\right)^2 + \left(\frac{1}{3}\right)^2 + k^2 = 1$
 i.e. $k^2 = 1 - \frac{1}{4} - \frac{1}{9} = \frac{23}{36}$
- 38.(a)** Here p is true, q is false.
 So $p \wedge q$ is false and $(p \wedge q) \Rightarrow r$ is true.
- 39.(c)** $\sin^{11}x \cos^{14}x$ is odd
 So, $\int_{-a}^a \sin^{11}x \cos^{14}x \, dx = 0$
- 40.(c)** $\lim_{x \rightarrow 0} \frac{\sin x}{x} = \lim_{x \rightarrow 0} \frac{\sin \frac{\pi x}{180}}{\frac{\pi x}{180}} \times \frac{\pi}{180} = \frac{\pi}{180}$
- 41.(a)** Put $y = \frac{1}{x}$
 So $dy = -\frac{1}{x^2} dx$
 Here $dx = 1 - 0.98 = -0.02$

Then $dy = -\frac{1}{12} \times -0.02 = 0.02$

42.(d) Here y-axis makes an angle of 90° with x-axis so
 $\text{slope} = \tan 90^\circ = \infty$

43.(b) Given $y = \log_{\sqrt{x}} x = 2 \log_x x = 2$

$\therefore \frac{dy}{dx} = 0$

44.(d) We have, $\int e^x \{f(x) + f'(x)\} dx = e^x f(x) + c$.

So, $f(x) = \tan x$ and hence $f'(x) = \sec^2 x$

45.(b) When a coin is tossed twice then $S = \{HH, HT, TH, TT\}$

So the required probability $= \frac{1}{4}$

46.(c) Given regression equation of y on x is

$3x + 5y = 15$

or, $5y = -3x + 15$

i.e. $y = -\frac{3}{5}x + 3$

So, $b_{yx} = -\frac{3}{5}$

47.(d) Given differential equation is $\frac{d^2y}{dx^2} = \sqrt[3]{1 + \left(\frac{dy}{dx}\right)^4}$

i.e. $\left(\frac{d^2y}{dx^2}\right)^3 = 1 + \left(\frac{dy}{dx}\right)^4$

So degree = 3

48.(b) $\int \log_e x dx = \int 1 \log_e x dx$

$= \log_e x \int dx - \int \left(\frac{d}{dx} \log_e x \int dx\right) dx$

So, $\int_1^e \log_e x dx = [x \log_e x - x]_1^e$
 $= (e \log_e e - e) - (1 \log 1 - 1)$
 $= (e - e) - (0 - 1) = 1$

49.b 50.c 51.b 52.b 53.a 54.c

55.d 56.d 57.b 58.b 59.a 60.b

Section – II

61.(a) KE will be least at maximum height where range $= \frac{R}{2}$

62.(c) $\frac{\left(\frac{d\theta}{dt}\right)_1}{\left(\frac{d\theta}{dt}\right)_2} = \frac{\theta_1 - \theta_0}{\theta_2 - \theta_0}$

$\frac{20}{\frac{7}{12}} = \frac{50 - 10}{34 - 10}$

or, $\frac{20t}{12 \times 7} = \frac{50 - 10}{34 - 10}$

or, $\frac{20t}{12 \times 7} = \frac{40}{24}$

or, $t = 7 \text{ min}$

63.(c) First case

$\frac{Q}{t} = \frac{KAd\theta}{l} = \frac{m_l L_f}{t}$

or, $\frac{KAd\theta}{l} = 0.1 \times L_f \dots (1)$

2nd case

$\frac{Q}{t} = \frac{K4Ad\theta \times 2}{4 \times l} = \frac{m_2}{t} L_f \dots (2)$

Dividing (2) by (1)

$\frac{m_2}{t} = 2 \times 0.1 = 0.2 \text{ g/s}$

64.(a) $f' = \frac{v + v_0}{v - v_s} \times f$

or, $\frac{f'}{f} = \frac{v + \frac{v}{2}}{v - \frac{v}{2}} = \frac{3v}{2} \times \frac{2}{v} = 3$

% increase $= \left(\frac{f'}{f} - 1\right) \times 100\%$
 $= (3 - 1) \times 100\%$
 $= 200\%$

65.(b) $\frac{2f}{f} = \sqrt{\frac{T}{g}}$

or, $T = 36 \text{ kg wt}$

66.(a) When one end of rod coincide with image then that end lies at C so

$u_1 = v_1 = 2f$

Diminished image means object lies beyond C

for end B

$u_2 = 2f + \frac{f}{2} = \frac{5f}{2}$

$v_2 = \frac{fu_2}{u_2 - f} = \frac{f \times \frac{5f}{2}}{\frac{5f}{2} - f} = \frac{5f^2}{2} \times \frac{2}{3f} = \frac{5f}{3}$

Length of image (l) $= V_1 - V_2 = 2f - \frac{5f}{3} = \frac{f}{3}$

67.(c) Total = 50 cm

In air distance = 40 cm

Real distance $= \frac{40}{\mu}$
 $= \frac{40}{4} \times 3 = 30 \text{ cm}$

$\therefore$ Actual distance $= 10 + 30 = 40 \text{ cm}$

68.(d) $0 = \sqrt{l_1 l_2} = \sqrt{4 \times 16} = 8 \text{ cm}$

69.(a) $\beta = \frac{D\lambda}{d}$

- $\Delta\beta = \frac{\lambda\Delta D}{d}$
 or, $\lambda = \frac{\Delta\beta \times d}{\Delta D} = \frac{3 \times 10^{-5} \times 10^{-3}}{5 \times 10^{-2}}$
 $= 6 \times 10^{-7} \text{ m} = 6000 \times 10^{-10} \text{ m}$
 $= 6000 \text{ \AA}$
- 70.(a)** At. surface of outer sphere
 $V = \frac{Q}{4\pi\epsilon_0 2R} + \frac{2Q}{4\pi\epsilon_0 2R}$
 $= \frac{3Q}{4\pi\epsilon_0 2R}$
 When sphere are connected by wire then all charge lies on outer sphere
 $V' = \frac{3Q}{4\pi\epsilon_0 2R} = V$
- 71.(c)** Total energy stored = energy stored in combination
 or, $\left(\frac{1}{2} \times 10^{-6} \times 500^2\right) n = \frac{1}{2} \times 10^{-6} (3000)^2$
 or, $n = \left(\frac{3000}{500}\right)^2 = 36$
- 72.(d)** Ladder are at same potential so act as open circuit so
 $R_{eq} = \frac{3R}{2}$
- 73.(d)** $\frac{N}{N_0} = 80\% = \left(\frac{1}{2}\right)^{\frac{t}{T_{1/2}}}$
 or, $\frac{4}{5} = \left(\frac{1}{2}\right)^{\frac{t}{T_{1/2}}}$
 or, $\ln\left(\frac{4}{5}\right) = \frac{t}{T_{1/2}} \ln\left(\frac{1}{2}\right)$
 or, $T_{1/2} = 31 \text{ days}$
 $\frac{N'}{N_0} = \left(\frac{1}{2}\right)^{\frac{t}{T_{1/2}}} = \left(\frac{1}{2}\right)^{\frac{20}{31}} = 0.639$
 $\% \text{ left} = \frac{N'}{N_0} \times 100\% = 0.639 \times 100\%$
 $= 63.9\%$
- 74.(b)** **Hint:** The chief ore of zinc is its sulphide. (Haematite is an iron ore, bauxite an aluminium ore, cinnabar a mercury ore.)
Solution: The chief ore of zinc is zinc blende (sphalerite), ZnS. It is roasted to ZnO and then reduced to obtain zinc.
- 75.(c)** **Hint:** Alkynes contain one carbon-carbon triple bond, so they have four fewer hydrogens than the corresponding alkane.
Solution: Alkynes have one C≡C triple bond and the general formula C_nH_{2n-2} (e.g. ethyne C_2H_2).
- 76.(b)** **Hint:** Ethene has a C=C double bond. Bromine adds across the double bond, so the reagent is consumed and decolourised.
- Solution:** Br₂ adds across the C=C double bond of ethene (electrophilic addition) to give 1,2-dibromoethane, decolourising the bromine water.
- 77.(c)** **Hint:** Freons are chlorofluorocarbons (CFCs) containing both chlorine and fluorine.
Solution: Dichlorodifluoromethane, CCl₂F₂ (Freon-12), is a chlorofluorocarbon used as a refrigerant.
- 78.(b)** **Hint:** Primary alcohols oxidise in two stages. Identify the first product before further oxidation to an acid.
Solution: A primary alcohol is first oxidised to an aldehyde, which on further oxidation gives a carboxylic acid. (Secondary alcohols give ketones.)
- 79.(a)** **Hint:** This is an esterification (Fischer esterification); concentrated H₂SO₄ acts as a catalyst and dehydrating agent.
Solution: Acetic acid + ethanol — (conc. H₂SO₄) → ethyl acetate (an ester) + water.
 $CH_3COOH + C_2H_5OH \rightarrow CH_3COOC_2H_5 + H_2O$
- 80.(c)**
 $2R - O - H \xrightarrow[170^\circ C]{\text{conc. H}_2\text{SO}_4} R - C \equiv C - R' + H_2O$
- 81.(b)**
82.(a) Here $n(A) + n(B) = 65 + 95 = 160 > 130 = n(U)$
 So maximum possible value of $n(A \cup B) = n(U) = 130$
- 83.(d)** Here $S_\infty = 15$, $a = 5$
 So $S_\infty = \frac{a}{1-r}$ or, $15 = \frac{5}{1-r}$ or, $1-r = \frac{5}{15}$
 or, $r = 1 - \frac{1}{3} \therefore r = \frac{2}{3}$
- 84.(c)** Given $z = r(\cos\theta + i\sin\theta)$
 So $iz = r(i\cos\theta - \sin\theta)$
 $= r \left[\cos\left(\frac{\pi}{2} + \theta\right) + i\sin\left(\frac{\pi}{2} + \theta\right) \right]$
 $\therefore \arg(iz) = \frac{\pi}{2} + \theta$
- 85.(a)** Given word is NUMBER
 For the arrangements that begin and end with vowel, U and E are kept fixed and remaining 4 letters are arranged in 4! ways. But U and E can interchange the position, so total no. of arrangements = $2 \times 4! = 48$.
- 86.(d)** Given series is $1 + \frac{a+bx}{1!} + \frac{(a+bx)^2}{2!} + \dots$
 $= e^{a+bx}$
 $= e^a e^{bx} = e^a \sum_{n=0}^{\infty} \frac{(bx)^n}{n!} = e^a \sum_{n=0}^{\infty} \frac{b^n x^n}{n!}$
 So, coefficient of $x^{10} = \frac{e^a b^{10}}{10!}$
- 87.(c)** Given, coefficient of x^r = coefficient of x^{r+1}
 or, $C(2n+1, r) = C(2n+1, r+1)$
 So, $r = r+1 \Rightarrow 1 = 0$, not possible

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2083-3-27 Hints & Solution

or, $r + r + 1 = 2n + 1 \Rightarrow 2r = 2n \therefore r = n$ 88.(b) Given $\frac{a}{\cos A} = \frac{b}{\cos B}$ or, $\frac{2R \sin A}{\cos A} = \frac{2R \sin B}{\cos B}$ or, $\tan A = \tan B$ or, $A = B$ $\therefore \Delta$ is isosceles 89.(b) Given circle is $x^2 + y^2 = 16$ $\therefore$ Length of tangent drawn from (4, 3) to the circle $\sqrt{x_1^2 + y_1^2 - 16} = \sqrt{4^2 + 3^2 - 16}$ $= \sqrt{16 + 9 - 16} = \sqrt{9} = 3 \text{ units}$ 90.(b) Given parabola is $y^2 = 16x$. So $a = 4$ and line is $y = 2x + c$ So $m = 2, c = c$ Thus line touches parabola if $c = \frac{a}{m} = \frac{4}{2} = 2$ 91.(d) Given $x = a \cos \theta, y = b \sin \theta$ So $\left(\frac{x}{a}\right)^2 + \left(\frac{y}{b}\right)^2 = \cos^2 \theta + \sin^2 \theta$ i.e. $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$, which an ellipse 92.(b) Given $x^2 - y^2 = a^2, xy = c^2$ which are rectangular hyperbola. So $e = \sqrt{2}, e' = \sqrt{2}$. Now, $e^2 + e'^2 = 2 + 2 = 4$	93.(a) Equation of plane parallel to $x + 2y + 2z + 7 = 0$ and passing through the point (1, 2, 2) is $(x - 1) + 2(y - 2) + 2(z - 2) = 0$ i.e. $x + 2y + 2z = 9$ 94.(c) Given $y = ax^2 - 12x + 20$ So $\frac{dy}{dx} = 2ax - 12$ As the function has local minimum value at $x = 2$. So $2a \times 2 - 12 = 0$ or, $4a = 12 \therefore a = 3$ 95.(b) Given $y = A \sin x + B \cos x$ So $\frac{dy}{dx} = A \cos x - B \sin x$ and $\frac{d^2y}{dx^2} = -A \sin x - B \cos x$ $= -(A \sin x + B \cos x) = -y$ $\therefore \frac{d^2y}{dx^2} + y = 0$ 96.(a) Given curve is $y = 2x - x^2$. Along x-axis, $y = 0$ So $2x - x^2 = 0$ i.e. $x = 0, 2$ $\therefore$ Required area $= \int_0^2 y dx = \int_0^2 (2x - x^2) dx$ $= \left[x^2 - \frac{x^3}{3} \right]_0^2 = \left(2^2 - \frac{2^3}{3} \right) - (0 - 0)$ $= 4 - \frac{8}{3} = \frac{4}{3}$ 97.c 98.d 99.b 100.a
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