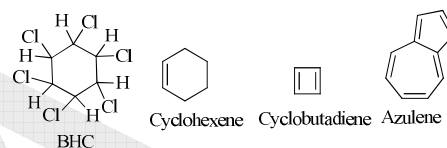


**Section – I**

- 1.(d)  $Bx = Dt$   
 or,  $\frac{D}{B} = LT^{-1}$
- 2.(b)  $R = \frac{u^2 \sin 2\theta}{g}$ , R will be max if  
 $\sin 2\theta = 1 = \sin 90^\circ$   
 $\theta = 45^\circ$
- 3.(c)  $\frac{1}{2} m (2v_e)^2 - \frac{1}{2} m v_e^2 = \frac{1}{2} m v'^2$   
 or,  $\sqrt{3} v_e = v'$
- 4.(c)  $r = \sqrt{r_1^2 + r_2^2} = \sqrt{3^2 + 4^2} = 5 \text{ cm}$
- 5.(c) i.e. 0 to  $4^\circ\text{C} \rightarrow$  volume decreases,  $4^\circ\text{C}$  to  $15^\circ\text{C} \rightarrow$  volume increases
- 6.(c)  $a_2 = 6 \text{ unit}, a_1 = 8 \text{ unit}$   
 $\frac{I_{\max}}{I_{\min}} = \left( \frac{a_1 + a_2}{a_1 - a_2} \right)^2 = \left( \frac{8+6}{8-6} \right)^2 = \left( \frac{14}{2} \right)^2 = 49 : 1$
- 7.(c)  $\sin 60^\circ = \frac{v_p}{v}$   
 $v_p = \frac{\sqrt{3}v}{2}$
- 8.(c)
- $$\frac{Q}{a/2} \cdot \frac{Q}{a/2} = \frac{Qq}{a^2} \cdot \frac{Q \cdot Q}{4\pi\epsilon_0 a^2} \rightarrow q = \frac{Q}{4}$$
- 9.(b)  $\frac{R'}{R} = \left( \frac{2l}{l} \right)^2 = 4$   
 $R' = 4R$
- 10.(c)  $E = \frac{\Delta\phi}{\Delta t} = \frac{8 \times 10^{-4}}{0.5} = 1.6 \text{ mV}$
- 11.(a)  $V_L = 60 \text{ V}, V_C = 30 \text{ V}, V_R = 40 \text{ V}$   
 $V = \sqrt{V_R^2 + (V_L - V_C)^2} = 50 \text{ V}$
- 12.(d)  $v = \sqrt{\frac{2eV}{m}} = \sqrt{2 \times 1.8 \times 10^{11} \times 100} = 6 \times 10^6 \text{ m/s}$
- 13.(a)  $\phi = hf_0 \Rightarrow f_0 = \frac{\phi_0}{h} = 8 \times 10^{14} \text{ Hz}$
- 14.(b) P-type semiconductor holes are the majority charge carriers.
- 15.(a) No. of protons = No. of mole  $\times N_A \times$  No. of protons in one molecule of  $\text{CaCO}_3$
- 16.(b)

- 17.(b)  $\text{MHPO}_4$  shows that valency of M = 2 (since  $\text{HPO}_4$  has valency 2). Hence chloride will be  $\text{MCl}_2$
- 18.(b)
- 19.(d)
- 20.(a)
- 21.(d)  $\text{F}^-$  is the most electronegative element.
- 22.(c) The impurity in extraction of copper is  $\text{FeO}$  which is removed by adding  $\text{SiO}_2$ .
- 23.(d) It obeys Huckel's rule i.e. it contains  $(4n+2)$  delocalized  $\pi$  electrons e.g. 10  $\pi$  electrons.



- 24.(a) It is known as enyne compound. Its IUPAC format is: Alk-en-yne. Numbering is done by the lowest sum rule.
- 25.(d) Carbonium ion e.g.  $\text{CH}_3^+$  (6 electrons)  
 Free radical e.g.  $\cdot\text{CH}_3$  (7 electrons)  
 Nitrene e.g.  $\text{CH}_3\text{N}$  (6 electrons)  
 Carbanion e.g.  $\text{CH}_3^-$  (8 electrons)
- 26.(d)  $(\text{CH}_3)_3\text{CNO}_2$ ,  $\text{CCl}_3\text{CHO}$  and  $(\text{CH}_3)_3\text{CHO}$  do not have  $\alpha$  hydrogen atoms so they do not show tautomerism.
- 27.(a) +R or +M groups viz.  $-\text{OH}$ ,  $\text{OR}$ ,  $-\text{NH}_2$ ,  $-\text{X}$  etc give ortho and para substituted product due to mesomeric effect or resonating effect.
- 28.(c)
- 29.(a)  $B \subset A$ , then  $A \cup B = A$
- 30.(a)  $z = \frac{1}{2+i} \times \frac{2-i}{2-i} = \frac{2-i}{2^2-i^2} = \frac{2-i}{5}$   
 $\bar{z} = \frac{2+i}{5}$
- 31.(c)  $AM \times H.M = GM^2$   
 or,  $H.M = \frac{G^2}{A}$
- 32.(c)  $\theta$  lies on 3<sup>rd</sup> quadrant.  
 $\therefore \theta = \pi + \frac{\pi}{6} = \frac{7\pi}{6}$   
 Most general value =  $2n\pi + \frac{7\pi}{6}$
- 33.(b) Focus =  $\left( \frac{-5+3}{2}, \frac{6+6}{2} \right) = (-1, 6)$
- 34.(b) Let  $\vec{a} = a_1\vec{i} + a_2\vec{j} + a_3\vec{k}$   
 Then,  $\vec{a} \cdot \vec{i} = a_1$ ,  $\vec{a} \cdot \vec{j} = a_2$ ,  $\vec{a} \cdot \vec{k} = a_3$   
 So,  $(\vec{a} \cdot \vec{i})\vec{i} + (\vec{a} \cdot \vec{j})\vec{j} + (\vec{a} \cdot \vec{k})\vec{k}$   
 $= a_1\vec{i} + a_2\vec{j} + a_3\vec{k} = \vec{a}$

35.(a)

36.(b) Let  $y = \sec^2 x$ ,  $z = \tan x$

$$\frac{dy}{dx} = 2 \sec x \cdot \sec x \tan x$$

$$\& \quad \frac{dz}{dx} = \sec^2 x$$

$$\therefore \frac{dy}{dz} = 2 \tan x$$

37.(b)  $\sin^{-1}(\cos x) = \frac{\pi}{2} - \cos^{-1}(\cos x)$

$$= \frac{\pi}{2} - x$$

$$\text{Now, } \int \sin^{-1}(\cos x) dx = \frac{\pi}{2} x - \frac{x^2}{2} + c$$

38.(c)  $xy = 1$  which is rectangular hyperbola. So,  $e = \sqrt{2}$ .

39.(a)

40.(b) (0, 1, 0)

41.(d) Greatest coefficient is the coeff. of mid term.

42.(d)  $f(x) = \frac{1}{3\sin x - 4\cos x + 7}$  will be minimum when  $3\sin x - 4\cos x + 7$  is maximum.

Maximum of denominator

$$= \sqrt{3^2 + 4^2} + 7 = 5 + 7 = 12$$

43.(c) The given equation are intersecting lines.

44.(c)  $ax + by = 2ab$

$$\text{or, } \frac{x}{2b} + \frac{y}{2a} = 1$$

$$A = \frac{1}{2} \times b \times h = \frac{1}{2} \cdot 2b \cdot 2a = 2ab$$

45.(a)  $\int_{-1}^2 |x| dx = \int_{-1}^0 (-x) dx + \int_0^2 x dx$

$$= \left[ -\frac{x^2}{2} \right]_{-1}^0 + \left[ \frac{x^2}{2} \right]_0^2$$

$$= -\left(0 - \frac{1}{2}\right) + \left(\frac{4}{2} - \frac{0}{2}\right)$$

$$= \frac{1}{2} + 2 = \frac{5}{2}$$

46.(b)

47.(a) Here,  $a = \cos^2 \theta - 1 = -\sin^2 \theta$

$$b = \sin^2 \theta$$

and  $a + b = 0$

So, the lines are perpendiculars

48.(a) Obvious

49.c    50.c    51.a    52.a    53.c    54.c  
 55.d    56.b    57.a    58.b    59.b    60.b

## Section – II

61.(c)  $\frac{h}{2} = \frac{g}{2} (2n - 1)$

$$\text{or, } \frac{1}{2} \times \frac{1}{2} g n^2 = \frac{g}{2} (2n - 1)$$

$$\text{or, } n^2 - 4n + 2 = 0$$

$$\text{or, } n = 3.42 \text{ sec}$$

$$\therefore h = \frac{1}{2} g (3.42)^2$$

$$= \frac{1}{2} \times 10 (3.42)^2$$

$$= 58 \text{ m}$$

62.(c)  $\frac{Gm_1}{x^2} = \frac{Gm_2}{(1-x)^2} \Rightarrow x = \frac{1}{11} \text{ m}$

63.(b) Energy stored = K.E. of mass

$$\frac{1}{2} \frac{Y A e^2}{m l} = \frac{1}{2} m v^2$$

$$v = \sqrt{\frac{Y A e^2}{m l}} = \sqrt{\frac{5 \times 10^8 \times 10^{-6} \times 0.02^2}{5 \times 10^{-3} \times 0.1}} = 20 \text{ m/s}$$

64.(b)  $E = \sigma A T^4 \times t = 4.45 \text{ kJ}$

65.(a)  $(\mu - 1)t = n\lambda, \lambda = \frac{(\mu - 1)t}{n} = \frac{(1.5 - 1) \times 6 \times 10^{-6}}{5}$

$$= 6 \times 10^{-7} \text{ m} = 6000 \text{ \AA}$$

66.(b)  $f_0 = \frac{1}{2L} \sqrt{\frac{\text{stress}}{\rho}} = \frac{1}{2l} \sqrt{\frac{Y \times \text{strain}}{\rho}} = 170 \text{ Hz}$

67.(c)  $F = 9 \times 10^9 \cdot \frac{Q_1 Q_2}{r^2} \Rightarrow r^2 = 9 \times 10^9 \frac{Q_1 Q_2}{F} = 9 \text{ cm}$

68.(b) Amount of heat energy required for the water to boil  
 $Q = 1 (100 - 20) \times 4200 + 420 \times 80 = 369600 \text{ J}$   
 $Q = 90\% \text{ of Pt, } t = 467 \text{ sec}$

69.(d)  $E = \frac{1}{2} m v^2$

$$v = \sqrt{\frac{2E}{m}} = \sqrt{\frac{2 \times 2 \times 10^6 \times 1.6 \times 10^{-19}}{1.67 \times 10^{-27}}}$$

$$= 1.96 \times 10^7 \text{ m/s}$$

$$F = Bev = 2.5 \times 1.6 \times 10^{-19} \times 1.96 \times 10^7$$

$$= 7.84 \times 10^{-12} \text{ N}$$

70.(c)  $L = \frac{N\phi}{I} = 2.5 \times 10^{-3} \text{ H}$

$$\text{The magnetic energy stored, } U = \frac{1}{2} L I^2 = 5 \times 10^{-3} \text{ J}$$

71.(d)  $\phi_0 = \frac{hc}{\lambda} - \text{K.E.} = 3 \times 10^{-19}$

$$f_0 = \frac{3 \times 10^{-19}}{h} = 4.5 \times 10^{14} \text{ Hz}$$

**72.(d)** For 1st member of Balmer series  $\frac{1}{\lambda_{B'}} = R \left( \frac{1}{4} - \frac{1}{9} \right)$

$$\Rightarrow \lambda_B = \frac{36}{5R}$$

For second member of same series,

$$\frac{1}{\lambda_{B'}} = R \left( \frac{1}{4} - \frac{1}{16} \right) \Rightarrow \lambda_{B'} = \frac{16}{3R} \dots\dots(i)$$

$$\text{i.e. } \frac{\lambda_{B'}}{\lambda_B} = \frac{16 \times 5R}{3R \times 36}$$

$$\therefore \lambda_{B'} = \frac{20}{27} \times 6563 = 4861 \text{ \AA}$$

**73.(b)**  $\frac{N'}{N_0}$

$$\text{So, } \frac{N}{N_0} = 1 - \frac{N'}{N_0} = 1 - \frac{1}{4} = \frac{3}{4}$$

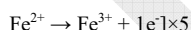
$$\frac{N}{N_0} = \left( \frac{1}{2} \right)^{\frac{t}{T_{1/2}}}$$

$$\text{or, } \frac{3}{4} = \left( \frac{1}{2} \right)^{\frac{t}{T_{1/2}}}$$

$$\text{or, } \frac{\ln\left(\frac{3}{4}\right)}{\ln\left(\frac{1}{2}\right)} = \frac{t}{T_{1/2}}$$

$$\text{or, } t = 224 \text{ yrs}$$

**74.(a)**  $\text{MnO}_4^- + 8\text{H}^+ + 5\text{e}^- \rightarrow \text{Mn}^{2+} + 4\text{H}_2\text{O}] \times 3$



As 5 moles of  $\text{Fe}(\text{C}_2\text{O}_4) = 3$  moles of  $\text{KmnO}_4$

So, 1 mol of  $\text{Fe}(\text{C}_2\text{O}_4) = 3/5$  moles of  $\text{KmnO}_4 = 0.6$  mol

**75.(a)**  $N_{\text{mix}} = (N_1V_1 + N_2V_2 + N_3V_3)/V_{\text{total}}$

**76.(b)** 71 parts of chlorine combine with 32 parts sulphur  
 35.5 parts of chlorine combine with 16 parts of sulphur

Hence, eq.wt of S in  $\text{SCl}_2 = 16$

**77.(d)** 1mol of Au = 197g = 0.197kg =  $6.02 \times 10^{23}$  atoms so,  
 19.7 kg Au =  $6.02 \times 10^{25}$  atoms

**78.(a)** No. of mol  $\times N_A$

**79.(c)** Bond length order: Single bond > bond created by resonance > double bond > triple bond

**80.(c)**

**81.(c)** B shows + I effect and hyperconjugation

C shows - I effect D shows -R and -I effect

$$\begin{aligned} \text{82.(a)} \quad \frac{dy}{dx} &= -\frac{fx}{fy} = -\frac{2ax + 2hy}{2hx + 2by} \\ &= -\frac{ax + hy}{hx + by} \end{aligned}$$

**83.(a)**  $f(x) = y = x^2 - 6x + 9 - 3$

$$y + 3 = (x - 3)^2 \geq 0$$

$$y + 3 \geq 0$$

$$y \geq -3$$

**84.(b)** z is a locus of a point whose distance from a point (3, 4) is always 5 unit. So, locus of z is a circle.

OR, put  $z = x + iy$  and solve.

$$\text{We get; } (x - 3)^2 + (y - 4)^2 = 25$$

**85.(a)** Since, one of the lines bisects the angle b/w the axes so the line is either  $y = x$  or

$$y = -x$$

Then the eq<sup>n</sup> is

$$ax^2 \pm 2hxy + by^2 = 0$$

$$\text{or, } a + b = \pm 2h$$

$$\therefore (a + b)^2 = 4h^2$$

**86.(d)** The line passes through the centre of the circle. So, it is a diameter.

$\therefore$  Angle between diameter and tangent of circle is  $90^\circ$

**87.(a)** If  $a = 0$ ,  $by + cz + d = 0$  is a plane parallel to x-axis.

$$\text{88.(b)} \quad \vec{a} \times \vec{b} = \begin{vmatrix} \vec{i} & \vec{j} & \vec{k} \\ 1 & 1 & 1 \\ -2 & 3 & 1 \end{vmatrix}$$

$$= -2\vec{i} - 3\vec{j} + 5\vec{k}$$

$$\text{Area} = \frac{1}{2} |\vec{a} \times \vec{b}| = \frac{\sqrt{38}}{2}$$

**89.(b)**  $a^{1/x} = b^{1/y} = c^{1/z} = k$

Then,  $a = k^x$ ,  $b = k^y$ ,  $c = k^z$

Since, a, b, c are in G.P.

$$b^2 = ac$$

$$k^{2y} = k^x \cdot k^z = k^{x+z}$$

$$\text{or, } 2y = x + z$$

∴ x, y, z are in A.P.

$$\begin{aligned} 90.(a) \quad \sum_{n=1}^{\infty} \frac{n^2}{n!} &= \sum_{n=1}^{\infty} \frac{n}{(n-1)!} \\ &= \sum_{n=1}^{\infty} \frac{n-1+1}{(n-1)!} \\ &= \sum_{n=1}^{\infty} \frac{1}{(n-2)!} + \sum_{n=1}^{\infty} \frac{1}{(n-1)!} \\ &= 2e \end{aligned}$$

$$\begin{aligned} 91.(c) \quad f'(x) &= 4x^3 + 12x^2 \\ f''(x) &= 12x^2 + 24x \\ &= 12x(x+2) \end{aligned}$$

Point of inflection,  $x = 0, x = -2$

i.e.  $x \in (-\infty, -2) \cup (0, \infty)$

$$92.(b) \quad \text{No of diagonals, } nC_2 - n = 144$$

It is true when  $n = 11$

$$\begin{aligned} 93.(d) \quad V &= \frac{4}{3} \pi r^3 \\ \frac{dv}{dt} &= \frac{4}{3} \pi 3r^2 \frac{dr}{dt} \\ 18 &= 4\pi r^2 \frac{dr}{dt} \end{aligned}$$

$$\frac{dr}{dt} = \frac{9}{128\pi} \text{ cm/sec}$$

$$94.(a) \quad \text{Let } y = \sin x$$

Then  $dy = \cos x \, dx$

When  $x = 0$ ;  $y = 0$  and when  $x = \frac{\pi}{2}$ ,  $y = 1$

$$\text{Then } \int_0^{\pi/2} \frac{\cos x}{1 + \sin^2 x} dx = \int_0^1 \frac{dy}{1 + y^2} = [\tan^{-1} y]_0^1 = \frac{\pi}{4}$$

$$\begin{aligned} 95.(c) \quad \text{Area} &= 2 \int_0^a y \, dx \\ &= 2\sqrt{4a} \int_0^a \frac{1}{2} dx \\ &= 4\sqrt{a} \cdot \frac{a^{3/2}}{3/2} = \frac{8}{3} a^2 \end{aligned}$$

$$\begin{aligned} 96.(c) \quad &\text{Apply, } R_2 \rightarrow R_2 - R_1 \\ &\text{and } R_3 \rightarrow R_3 - R_1 \\ \Delta &= \begin{vmatrix} 1 & 1 & 1 \\ 0 & x & 0 \\ 0 & 0 & y \end{vmatrix} = xy \\ \text{i.e. } \Delta &\text{ is divisible by both } x \text{ \& } y \end{aligned}$$

$$97.(c) \quad \begin{matrix} 98.(b) & 99.(c) & 100.(d) \end{matrix}$$

...The End...